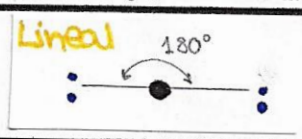
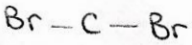
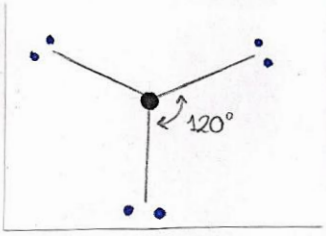
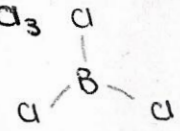
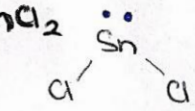
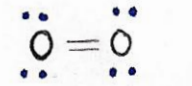
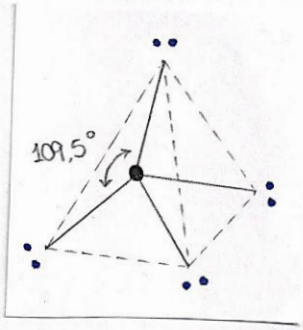
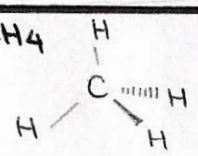
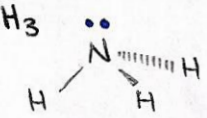
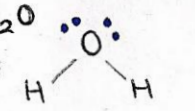
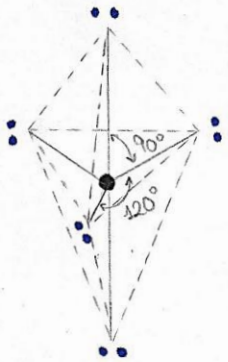
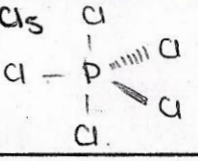
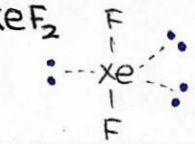
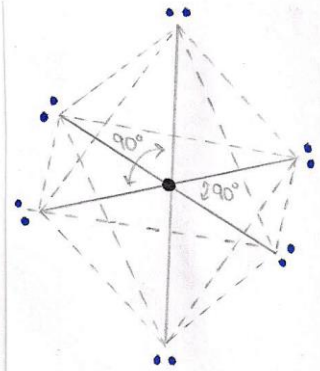
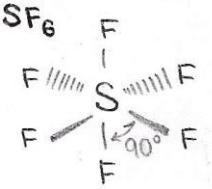
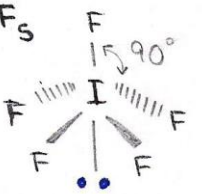
HIBRIDACIÓN	ORDENAMIENTO GENERAL	PARES DE ELECTRONES.	GEO. MOLECULAR.	EJEMPLO.
Sp (2 zonas)	Lineal 	2 pares enlace.	Lineal (Linear)	$CdBr_2$ 
Sp^2 (3 zonas)	Triangular plana. 	3 pares enlace.	Triangular plana. (Trigonal planar)	BCl_3 
		2 pares enlace. + 1 par solitario.	Angular (Angular/bent)	$SnCl_2$ 
		1 par enlace. + 2 pares solitarios.	Lineal (Linear)	O_2 
Sp^3 (4 zonas)	Tetrahedro. 	4 pares enlace.	Tetrahedro regular. (Tetrahedral)	CH_4 
		3 pares enlace. + 1 par solitario.	Pirámide trigonal. (Trigonal pyramid)	NH_3 
		2 pares enlace. + 2 pares solitarios.	Angular (Angular/bent)	H_2O 
		1 par enlace. + 3 pares solitarios.	Lineal (Linear.)	OH^- 
Sp^3d. (5 zonas)	Bipirámide trigonal. 	5 pares enlace.	Bipirámide trigonal. (Trigonal bipyramidal.)	PCl_5 
		4 pares enlace. + 1 par solitario.	Tetrahedro distorsionado. (Disated tetrahedral/seesaw)	SF_4 
		3 pares enlace. + 2 pares solitarios.	Forma de T. (T-Shaped / trigonal)	ClF_3 
		2 pares enlace. + 3 pares solitarios	Lineal (Linear.)	XeF_2 

sp^3d^2 (6 zonas)	Octaedro. 	6 pares enlace.	Octaedro (octahedral.)	SF_6 
		5 pares enlace + 1 par solitario.	Pirámide cuadrada. (square pyramidal.)	IF_5 
		4 pares enlace. + 2 pares solitarios	Cuadrado plana (square planar)	XeF_4